

VANDERBILT UNIVERSITY
MATH 196 — EXAMPLES OF SECTION 1.4.

Question 1. The intensity I of the light at a depth of x meters below the surface of a lake satisfies the differential equations $I' = -1.4I$.

- (a) At what depth is the intensity half of the intensity I_0 at the surface of the water?
- (b) What is the intensity at a depth of 10 meters?
- (c) At what depth will the intensity be 1 % of that at the surface?

Question 2. According to one cosmological theory, there were equal amounts of the two uranium isotopes ^{235}U and ^{238}U at the creation of the universe in the big bang. At present there are 137.7 atoms of ^{238}U for each atom of ^{235}U . Using the half-lives 4.51×10^9 years for ^{238}U and 7.10×10^8 years for ^{235}U , calculate the age of the universe.

SOLUTIONS.

1a. The differential equation is of the form $x' = kx$, which we saw in class that has solution $x(t) = x_0e^{kt}$, hence the intensity at a depth of x meters is $I(x) = I_0e^{-1.4x}$. Then

$$I(x) = \frac{I_0}{2} = I_0e^{-1.4x} \Rightarrow x = \frac{\ln 2}{1.4} \approx 0.495 \text{ meters.}$$

1b. Plugging in, $I(10) = I_0e^{-1.4 \times 10} \approx 8.3 \times 10^{-7}$.

1c. Solving $I_0e^{-1.4x} = 0.01I_0$ for x gives $x = \frac{\ln 100}{1.4} \approx 3.29$ meters.

2. Let $N_8(t)$ and $N_5(t)$ be the numbers of ^{238}U and ^{235}U atoms, respectively, t billions of years after the big bang. Since both isotopes follow a radioactive decay model $x' = kx$, whose solution was seen in class to be $x(t) = x_0e^{kt}$, we have

$$N_8 = N_0e^{-kt},$$

and

$$N_5 = N_0e^{-\ell t},$$

where N_0 is the initial number of atoms of each isotope, which is the same for both ^{238}U and ^{235}U by hypothesis. Notice however that the rates of decay, k and ℓ , differ for these isotopes. Their values are given by

$$\begin{aligned} N_8(4.51) &= \frac{N_0}{2} = N_0e^{-k \times 4.51} \Rightarrow k = \frac{\ln 2}{4.51}, \\ N_5(0.71) &= \frac{N_0}{2} = N_0e^{-\ell \times 0.71} \Rightarrow \ell = \frac{\ln 2}{0.71}. \end{aligned}$$

We know that for the value of t corresponding to “now” we have $\frac{N_8}{N_5} = 137.7$, hence

$$\frac{N_8}{N_5} = \frac{N_0e^{-kt}}{N_0e^{-\ell t}} = e^{(\ell-k)t} = e^{\left(\frac{\ln 2}{0.71} - \frac{\ln 2}{4.51}\right)t} = 137.7.$$

Solving for t gives

$$t = \frac{\ln 137.7}{\frac{\ln 2}{0.71} - \frac{\ln 2}{4.51}} \approx 5.99.$$

According to this theory, therefore, the universe should be about 6 billion years old.

Note: According to our best current models, the age of the universe is estimated to be about 13.7 billions of years, and the initial ratio of ^{235}U to ^{238}U is estimated to be 1.65 rather than one, as in the exercise¹. See S. Weinberg, *Cosmology*, Oxford University Press. The interested student can consult the non-technical book *The First Three Minutes: A Modern View Of The Origin Of The Universe*, by the same author.

URL: <http://www.disconzi.net/Teaching/MAT196-Spring-15/MAT196-Spring-15.html>

¹It makes sense that it is larger than one because three additional neutrons must be added to the progenitor of ^{235}U to form the progenitor of ^{238}U .