VANDERBILT UNIVERSITY

MATH 2420 – METHODS OF ORDINARY DIFFERENTIAL EQUATIONS

Examples of section 2.2

Question 1. The intensity I of the light at a depth of x meters below the surface of a lake satisfies the differential equations I' = -1.4I.

- (a) At what depth is the intensity half of the intensity I_0 at the surface of the water?
- (b) At what depth will the intensity be 1 % of that at the surface?

Question 2. According to one cosmological theory, there were equal amounts of the two uranium isotopes ^{235}U and ^{238}U at the creation of the universe in the big bang. At present there are 137.7 atoms of ^{238}U for each atom of ^{235}U . Using the half-lives 4.51×10^9 years for ^{238}U and 7.10×10^8 years for ^{235}U , and assuming a radioactive decay model of the form x' = kx, calculate the age of the universe.

SOLUTIONS.

1a. The differential equation is of the form x' = kx. This equation is separable, and can be written as

$$\frac{dx}{x} = k \, dt,$$

provided that $x \neq 0$. Integrating

$$\int \frac{dx}{x} = k \int dt \Rightarrow \ln|x| = kt + C,$$

where C is an arbitrary constant of integration. Thus,

$$|x| = e^C e^{kt}.$$

We can remove the absolute value by introducing a sign, i.e., the above tells us that

$$x = \pm e^C e^{kt}.$$

Since C is an arbitrary constant, so is e^C , and therefore we can set $A = \pm e^C$ for some constant A. Our solution then reads

$$x(t) = Ae^{kt}.$$

Next, we notice that the constant A has a very simple interpretation: if x_0 is the value of x at t = 0, i.e., $x(0) = x_0$, then

$$x(0) = x_0 = Ae^{k0} \Rightarrow A = x_0,$$

and therefore we can write the solution as

$$x(t) = x_0 e^{kt}.$$

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Now we turn our attention to the problem in question. We have that the intensity at a depth of x meters is $I(x) = I_0 e^{-1.4x}$. Then

$$I(x) = \frac{I_0}{2} = I_0 e^{-1.4x} \Rightarrow x = \frac{\ln 2}{1.4} \approx 0.495 \, meters.$$

1b. Solving $I_0 e^{-1.4x} = 0.01 I_0$ for x gives $x = \frac{\ln 100}{1.4} \approx 3.29$ meters.

2. Let $N_8(t)$ and $N_5(t)$ be the numbers of ^{238}U and ^{235}U atoms, respectively, t billions of years after the big bang. Since both isotopes follow a radioactive decay model x' = kx, whose solution is (see previous problem) $x(t) = x_0 e^{kt}$, we have

$$N_8 = N_0 e^{-kt},$$

and

$$N_5 = N_0 e^{-\ell t}$$

where N_0 is the initial number of atoms of each isotope, which is the same for both ^{238}U and ^{235}U by hypothesis. Notice however that the rates of decay, k and ℓ , differ for these isotopes. Their values are given by

$$N_8(4.51) = \frac{N_0}{2} = N_0 e^{-k \times 4.51} \Rightarrow k = \frac{\ln 2}{4.51},$$

$$N_5(0.71) = \frac{N_0}{2} = N_0 e^{-\ell \times 0.71} \Rightarrow \ell = \frac{\ln 2}{0.71}.$$

We know that for the value of t corresponding to "now" we have $\frac{N_8}{N_5}=137.7$, hence

$$\frac{N_8}{N_5} = \frac{N_0 e^{-kt}}{N_0 e^{-\ell t}} = e^{(\ell - k)t} = e^{(\frac{\ln 2}{0.71} - \frac{\ln 2}{4.51})t} = 137.7.$$

Solving for t gives

$$t = \frac{\ln 137.7}{\frac{\ln 2}{0.71} - \frac{\ln 2}{4.51}} \approx 5.99.$$

According to this theory, therefore, the universe should be about 6 billion years old.

Note: According to our best current models, the age of the universe is estimated to be about 13.7 billions of years, and the initial ratio of ^{235}U to ^{238}U is estimated to be 1.65 rather than one, as in the exercise¹. See S. Weinberg, Cosmology, Oxford University Press. The interested student can consult the non-technical book $The\ First\ Three\ Minutes:\ A\ Modern\ View\ Of\ The\ Origin\ Of\ The\ Universe$, by the same author.

¹It makes sense that it is larger than one because three additional neutrons must be added to the progenitor of ^{235}U to from the progenitor of ^{238}U .